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- Research Article-Chemistry
 - [Published: 09 June 2022](#)

Exploration of the Corrosion Inhibition Potential of Cashew Nutshell on Thermo-Mechanically Treated Steel in Seawater

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[Arabian Journal for Science and Engineering](#) **volume 48**, pages223–237
(2023)[Cite this article](#)

Abstract

This study explores the non-edible part of Cashew nutshell for possible conversion to useful product as a corrosion inhibitor for thermo-mechanically treated steel (TMT) in seawater. Herein, methanolic Cashew nutshell extract is examined as a corrosion inhibitor for TMT in seawater. The extract is rich in compounds such as saponins, phenol, and terpenoids according to results obtained from phytochemical analysis. The inhibitive performance of the extract was assessed using the gravimetric, spectroscopic, and electrochemical methods complemented with scanning electron microscope studies. The extract is effective in controlling the corrosion of TMT in seawater with 500 ppm capable of exerting over 75% inhibition efficiency at 30 °C for as long as 15 days. Results from spectroscopic studies [ultraviolet–visible spectroscopy

(UV–Vis) and Fourier transform infrared spectroscopy (FTIR)] reveal that the extract inhibits the TMT corrosion by adsorptive mechanism using the oxygen atoms and the aromatic rings that characterized the extract's components. However, the protection efficiency of the extract slightly depreciated with rise in the temperature of the corrosive medium. The surface morphological studies show evidence of extract components adsorption on the TMT surface, and the FTIR results confirmed the observation. The Cashew nutshell extract could be a sustainable source for the formation of an effective corrosion inhibitor for TMT, which is a common engineering material in seawater environment.

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References

-
1. Hou, B.; Li, X.; Ma, X.; Du, C.; Zhang, D.; Zheng, M.; Xu, W.; Lu, D.; Ma, F.: The cost of corrosion in China. NPJ Mater. Degrad. (2017). <https://doi.org/10.1038/s41529-017-0005-2>

[Article Google Scholar](#)

2. Farhan, N.A.; Sheikh, M.N.; Hadi, M.N.S.: Experimental investigation on the Effect of corrosion on the bond between reinforcing steel bars and fibre reinforced geopolymer concrete. Structures **14**, 251–261 (2018). <https://doi.org/10.1016/j.istruc.2018.03.013>
-

[Article Google Scholar](#)

3. Niu, D.; Zhang, L.; Fu, Q.; Wen, B.; Luo, D.: Critical conditions and life prediction of reinforcement corrosion in coral aggregate concrete. *Constr. Build. Mater.* **238**, 117685 (2020). <https://doi.org/10.1016/j.conbuildmat.2019.117685>

[Article Google Scholar](#)

4. IgibahEhizemhen, C.; Agashua Lucia, O.; SadiqAbubakar, A.: Statistical analysis and corrosion assessment of Nigeria steel rebars: case study south-west. Nigeria. *Civ. Eng. Archit.* **7**, 278–285 (2019). <https://doi.org/10.13189/cea.2019.070602>

[Article Google Scholar](#)

5. Hansapinyo, C.; Vimonsatit, V.; Matsushima, M.; Limkatanyu, S.: Critical amount of corrosion and failure behavior of flexural reinforced concrete beams. *Constr. Build. Mater.* **270**, 121448 (2021). <https://doi.org/10.1016/j.conbuildmat.2020.121448>

[Article Google Scholar](#)

6. Shaw, B.A.; Kelly, R.G.: What is corrosion? *Electrochem. Soc. Interface.* **15**, 24–26 (2006). <https://doi.org/10.1149/2.f06061if>

[Article Google Scholar](#)

7. Obike, A.I.; Uwakwe, K.J.; Abraham, E.K.; Ikeuba, A.I.; Emori, W.: Review of the losses and devastation caused by corrosion in the Nigeria oil industry for over 30 years. *Int. J. Corros. Scale Inhib.* **9**, 74–91 (2020). <https://doi.org/10.17675/2305-6894-2020-9-1-5>

[Article Google Scholar](#)

8. Loto, R.T.; Loto, C.A.; Popoola, A.P.: Inhibition effect of deanol on mild steel corrosion in dilute sulphuric acid. *S. Afr. J. Chem.* **68**, 105–114 (2015). <https://doi.org/10.17159/0379-4350/2015/v68a16>
-

[Article Google Scholar](#)

9. Liu, Q.; Song, Z.; Han, H.; Donkor, S.; Jiang, L.; Wang, W.; Chu, H.: A novel green reinforcement corrosion inhibitor extracted from waste *Platanus acerifolia* leaves. *Constr. Build. Mater.* **260**, 119695 (2020). <https://doi.org/10.1016/j.conbuildmat.2020.119695>
-

[Article Google Scholar](#)

10. Umoren, S.A.; AlAhmary, A.A.; Gasem, Z.M.; Solomon, M.M.: Evaluation of chitosan and carboxymethyl cellulose as ecofriendly corrosion inhibitors for steel. *Int. J. Biol. Macromol.* **117**, 1017–1028 (2018). <https://doi.org/10.1016/j.ijbiomac.2018.06.014>
-

[Article Google Scholar](#)

11. Solomon, M.M.; Umoren, S.A.; Obot, I.B.; Sorour, A.A.; Gerengi, H.: Exploration of dextran for application as corrosion inhibitor for steel in strong acid environment: effect of molecular weight, modification, and temperature on efficiency. *ACS Appl. Mater. Interfaces* **10**, 28112–28129 (2018). <https://doi.org/10.1021/acsami.8b09487>
-

[Article Google Scholar](#)

12. Zuo, X.; Li, W.; Luo, W.; Zhang, X.; Qiang, Y.; Zhang, J.; Li, H.; Tan, B.: Research of *Lilium brownii* leaves extract as a commendable and green inhibitor for X70 steel corrosion in hydrochloric acid. *J. Mol. Liq.* **321**, 114914 (2021). <https://doi.org/10.1016/j.molliq.2020.114914>
-

[Article Google Scholar](#)

13. Thanh, L.T.; Vu, N.S.H.; Binh, P.M.Q.; Dao, V.A.; Thu, V.T.H.; Van Hien, P.; Panaitescu, C.; Nam, N.D.: Combined experimental and computational

studies on corrosion inhibition of *Houttuynia cordata* leaf extract for steel in HCl medium. J. Mol. Liq. **315**, 113787 (2020). <https://doi.org/10.1016/j.molliq.2020.113787>

[Article Google Scholar](#)

14. Palanisamy, S.P.; Maheswaran, G.; Selvarani, A.G.; Kamal, C.; Venkatesh, G.: Ricinus communis—a green extract for the improvement of anti-corrosion and mechanical properties of reinforcing steel in concrete in chloride media. J. Build. Eng. **19**, 376–383 (2018). <https://doi.org/10.1016/j.jobbe.2018.05.020>
-

[Article Google Scholar](#)

15. Tan, B.; Xiang, B.; Zhang, S.; Qiang, Y.; Xu, L.; Chen, S.; He, J.: Papaya leaves extract as a novel eco-friendly corrosion inhibitor for Cu in H₂SO₄ medium. J. Colloid Interface Sci. **582**, 918–931 (2021). <https://doi.org/10.1016/j.jcis.2020.08.093>
-

[Article Google Scholar](#)

16. Loto, C.A.; Loto, R.T.: Corrosion inhibition effect of Allium Cepa extracts on mild steel in H₂SO₄. Der Pharma Chem. **8**, 272–281 (2016)
-

[Google Scholar](#)

17. Umoren, S.A.; Solomon, M.M.; Obot, I.B.; Suleiman, R.K.: Date palm leaves extract as a green and sustainable corrosion inhibitor for low carbon steel in 15 wt.% HCl solution: the role of extraction solvent on inhibition effect. Environ. Sci. Pollut. Res. **28**, 40879–40894 (2021). <https://doi.org/10.1007/s11356-021-13567-5>
-

[Article Google Scholar](#)

18. Nigerian Finder: Cashew Farming in Nigeria: How to Start in 2019. <https://nigerianfinder.com/cashew-farming-in-nigeria/>
-

19. Vedharaj, S.; Vallinayagam, R.; Yang, W.M.; Chou, S.K.; Chua, K.J.E.; Lee, P.S.: Experimental and finite element analysis of a coated diesel engine fueled by cashew nut shell liquid biodiesel. *Exp. Therm. Fluid Sci.* **53**, 259–268 (2014). <https://doi.org/10.1016/j.expthermflusci.2013.12.018>
-

[Article Google Scholar](#)

20. Oliveira, M.S.C.; de Moraes, S.M.; Magalhães, D.V.; Batista, W.P.; Vieira, Í.G.P.; Craveiro, A.A.; de Manazes, J.E.S.A.; Carvalho, A.F.U.; de Lima, G.P.G.: Antioxidant, larvicidal and antiacetylcholinesterase activities of cashew nut shell liquid constituents. *Acta Trop.* **117**, 165–170 (2011). <https://doi.org/10.1016/j.actatropica.2010.08.003>
-

[Article Google Scholar](#)

21. Tippayawong, N.; Chaichana, C.; Promwangkwa, A.; Rerkkriangkrai, P.: Gasification of cashew nut shells for thermal application in local food processing factory. *Energy Sustain. Dev.* **15**, 69–72 (2011). <https://doi.org/10.1016/j.esd.2010.10.001>
-

[Article Google Scholar](#)

22. Furtado, L.B.; Nascimento, R.C.; Seidl, P.R.; Guimarães, M.J.O.C.; Costa, L.M.; Rocha, J.C.; Ponciano, J.A.C.P.: Eco-friendly corrosion inhibitors based on Cashew nut shell liquid (CNSL) for acidizing fluids. *J. Mol. Liq.* **284**, 393–404 (2019)
-

[Article Google Scholar](#)

23. Das, S.; Thalukdar, S.; Mukhopadhyay, G.; Bhattacharya, S.: Failure analysis of thermo-mechanically treated (TMT) bar during bending operation: a metallurgical investigation. *Trends Civ. Eng. its Archit.* **2**, 332–335 (2018). <https://doi.org/10.32474/TCEIA.2018.02.000150>
-

[Article Google Scholar](#)

24. Islam, M.A.: Corrosion behaviours of high strength TMT steel bars for reinforcing cement concrete structures. *Procedia Eng.* **125**, 623–630 (2015). <https://doi.org/10.1016/J.PROENG.2015.11.084>

[Article Google Scholar](#)

25. ASTM G31–12a: Standard Guide for Laboratory Immersion Corrosion Testing of Metals. Annu. B. ASTM Stand. (2010). <https://doi.org/10.1520/G0031-12A>

26. Umoren, S.A.; Obot, I.B.; Israel, A.U.; Asuquo, P.O.; Solomon, M.M.; Eduok, U.M.; Udoh, A.P.: Inhibition of mild steel corrosion in acidic medium using coconut coir dust extracted from water and methanol as solvents. *J. Ind. Eng. Chem.* **20**, 3612–3622 (2014). <https://doi.org/10.1016/j.jiec.2013.12.056>

[Article Google Scholar](#)

27. Oguzie, E.E.: Evaluation of the inhibitive effect of some plant extracts on the acid corrosion of mild steel. *Corros. Sci.* **50**, 2993–2998 (2008). <https://doi.org/10.1016/J.CORSCI.2008.08.004>

[Article Google Scholar](#)

28. Ahanotu, C.C.; Onyechu, I.B.; Solomon, M.M.; Chikwe, I.S.; Chikwe, O.B.; Eziukwu, C.A.: Pterocarpus santalinoides leaves extract as a sustainable and potent inhibitor for low carbon steel in a simulated pickling medium. *Sustain. Chem. Pharm.* **15**, 100196 (2020). <https://doi.org/10.1016/j.scp.2019.100196>

[Article Google Scholar](#)

29. Barua, A.G.; Hazarika, S.; Hussain, M.; Misra, A.K.: Spectroscopic investigation of the cashew nut kernel (*Anacardium occidentale*) spectroscopic investigation of the cashew nut kernel (*Anacardium occidentale*). *Open Food Sci. J.* **2**, 85–88 (2008). <https://doi.org/10.2174/1874256400802010085>

[Article Google Scholar](#)

30. Kyei, S.K.; Akaranta, O.; Darko, G.; Chukwu, U.J.: Extraction, characterization and application of cashew nut shell liquid from cashew nut shells. *Chem. Sci. Int. J.* **28**, 1–10 (2019). <https://doi.org/10.9734/CSJI/2019/v28i330143>
-

[Article Google Scholar](#)

31. Nandiyanto, A.B.D.; Oktiani, R.; Ragadhita, R.: How to read and interpret FTIR spectroscopy of organic material. *Indones. J. Sci. Technol.* **4**, 97–118 (2019). <https://doi.org/10.17509/IJOST.V4I1.15806>
-

[Article Google Scholar](#)

32. Sushmitha, Y.; Rao, P.: Electrochemical investigation on the acid corrosion control of mild steel using biopolymer as an inhibitor. *Port. Electrochim. Acta.* **38**, 149–163 (2020). <https://doi.org/10.4152/pea.202003149>
-

[Article Google Scholar](#)

33. Yaocheng, Y.; Caihong, Y.; Singh, A.; Lin, Y.: Electrochemical study of commercial and synthesized green corrosion inhibitors for N80 steel in acidic liquid. *New J. Chem.* **43**, 16058–16070 (2019). <https://doi.org/10.1039/c9nj03378e>
-

[Article Google Scholar](#)

34. Zhang, W.; Ma, R.; Liu, H.; Liu, Y.; Li, S.; Niu, L.: Electrochemical and surface analysis studies of 2-(quinolin-2-yl)quinazolin-4(3H)-one as corrosion inhibitor for Q235 steel in hydrochloric acid. *J. Mol. Liq.* **222**, 671–679 (2016). <https://doi.org/10.1016/j.molliq.2016.07.119>
-

[Article Google Scholar](#)

35. Boudellioua, H.; Hamlaoui, Y.; Tifouti, L.; Pedraza, F.: Effects of polyethylene glycol (PEG) on the corrosion inhibition of mild steel by cerium nitrate in chloride solution. *Appl. Surf. Sci.* **473**, 449–460 (2018). <https://doi.org/10.1016/j.apsusc.2018.12.164>

[Article Google Scholar](#)

36. Fiori-bimbi, M.V.; Alvarez, P.E.; Vaca, H.; Gervasi, C.A.: Corrosion inhibition of mild steel in HCL solution by pectin. *Corros. Sci.* **92**, 192–199 (2015). <https://doi.org/10.1016/j.corsci.2014.12.002>

[Article Google Scholar](#)

37. Palomar-Pardavé, M.; Romero-Romo, M.; Herrera-Hernández, H.; Abreu-Quijano, M.A.; Likhanova, N.V.; Uruchurtu, J.; Juárez-García, J.M.: Influence of the alkyl chain length of 2 amino 5 alkyl 1, 3, 4 thiadiazole compounds on the corrosion inhibition of steel immersed in sulfuric acid solutions. *Corros. Sci.* **54**, 231–243 (2012)

[Article Google Scholar](#)

38. Feliu, S.: Electrochemical impedance spectroscopy for the measurement of the corrosion rate of magnesium alloys: brief review and challenges. *Metals (Basel)* **10**, 775 (2020)

[Article Google Scholar](#)

39. Migahed, M.A.; Mohamed, H.M.; Al-Sabagh, A.M.: Corrosion inhibition of H-11 type carbon steel in 1 M hydrochloric acid solution by N-propyl amino lauryl amide and its ethoxylated derivatives. *Mater. Chem. Phys.* **80**, 169–175 (2003)

[Article Google Scholar](#)

40. Du, Y.T.; Wang, H.L.; Chen, Y.R.; Qi, H.P.; Jiang, W.F.: Synthesis of baicalin derivatives as eco-friendly green corrosion inhibitors for aluminum in

hydrochloric acid solution. J. Environ. Chem. Eng. **5**, 5891–5901 (2017). <https://doi.org/10.1016/j.jece.2017.11.004>

[Article Google Scholar](#)

41. Umoren, S.A.; Solomon, M.M.; Obot, I.B.: Effect of intensifier additives on the performance of butanolic extract of date palm leaves against the corrosion of API 5L X60 carbon steel in 15wt% HCl solution. Sustainability **13**, 5569 (2021)
-

[Article Google Scholar](#)

42. Paul, P.K.; Yadav, M.: Investigation on corrosion inhibition and adsorption mechanism of triazine-thiourea derivatives at mild steel/HCl solution interface: electrochemical, XPS, DFT and Monte Carlo simulation approach. J. Electroanal. Chem. **877**, 114599 (2020). <https://doi.org/10.1016/j.jelechem.2020.114599>
-

[Article Google Scholar](#)

43. Gerengi, H.; Mielniczek, M.; Gece, G.; Solomon, M.M.: Experimental and quantum chemical evaluation of 8-hydroxyquinoline as a corrosion inhibitor for copper in 0.1 M HCl. Ind. Eng. Chem. Res. (2016). <https://doi.org/10.1021/acs.iecr.6b02414>
-

[Article Google Scholar](#)

44. Gerengi, H.; Uygur, I.; Solomon, M.; Yildiz, M.; Goksu, H.: Evaluation of the inhibitive effect of Diospyros kaki (Persimmon) leaves extract on St37 steel corrosion in acid medium. Sustain. Chem. Pharm. (2016). <https://doi.org/10.1016/j.scp.2016.10.003>
-

[Article Google Scholar](#)

45. Salarvand, Z.; Amirnasr, M.; Talebian, M.; Raeissi, K.; Meghdadi, S.: Enhanced corrosion resistance of mild steel in 1 M HCl solution by trace amount of 2-phenyl-benzothiazole derivatives: experimental, quantum

chemical calculations and molecular dynamics (MD) simulation studies. *Corros. Sci.* **114**, 133–145 (2017). <https://doi.org/10.1016/J.CORSCI.2016.11.002>

[Article Google Scholar](#)

46. Fouda, A.S.; Abousalem, A.S.; El-Ewady, G.Y.: Mitigation of corrosion of carbon steel in acidic solutions using an aqueous extract of *Tilia cordata* as green corrosion inhibitor. *Int. J. Ind. Chem.* **8**, 61–73 (2017). <https://doi.org/10.1007/s40090-016-0102-z>
-

[Article Google Scholar](#)

47. Obot, I.B.; Onyeachu, I.B.: Electrochemical frequency modulation (EFM) technique: theory and recent practical applications in corrosion research (2018)
48. Verma, D.K.; Aslam, R.; Aslam, J.; Quraishi, M.A.; Ebenso, E.E.; Verma, C.: Computational modeling: theoretical predictive tools for designing of potential organic corrosion inhibitors. *J. Mol. Struct.* **1236**, 130294 (2021). <https://doi.org/10.1016/J.MOLSTRUC.2021.130294>
-

[Article Google Scholar](#)

49. Saxena, A.; Prasad, D.; Haldhar, R.: Investigation of corrosion inhibition effect and adsorption activities of *Cuscuta reflexa* extract for mild steel in 0.5 M H₂SO₄. *Bioelectrochemistry* **124**, 156–164 (2018). <https://doi.org/10.1016/j.bioelechem.2018.07.006>
-

[Article Google Scholar](#)

50. Mohammed, A.R.I.; Solomon, M.M.; Haruna, K.; Umoren, S.A.; Saleh, T.A.: Evaluation of the corrosion inhibition efficacy of *Cola acuminata* extract for low carbon steel in simulated acid pickling environment. *Environ. Sci. Pollut. Res.* **27**, 34270–34288 (2020). <https://doi.org/10.1007/s11356-020-09636-w>
-

[Article Google Scholar](#)

51. Benea, L.; Simionescu, N.; Mardare, L.: The effect of polymeric protective layers and the immersion time on the corrosion behavior of naval steel in natural seawater. *J. Mater. Res. Technol.* **9**, 13174–13184 (2020). <https://doi.org/10.1016/j.jmrt.2020.09.059>
-

[Article Google Scholar](#)

52. Obot, I.B.; Solomon, M.M.; Onyeachu, I.B.; Umoren, S.A.; Meroufel, A.; Alenazi, A.; Sorour, A.A.: Development of a green corrosion inhibitor for use in acid cleaning of MSF desalination plant. *Desalination* **495**, 114675 (2020). <https://doi.org/10.1016/j.desal.2020.114675>
-

[Article Google Scholar](#)

53. Ebenso, E.E.; Ekpe, U.J.; Umoren, S.A.; Jackson, E.; Abiola, O.K.; Oforka, N.C.: Synergistic effect of halide ions on the corrosion inhibition of aluminum in acidic medium by some polymers. *J. Appl. Polym. Sci.* **100**, 2889–2894 (2006). <https://doi.org/10.1002/app.23505>
-

[Article Google Scholar](#)

54. Umoren, S.A.; Ebenso, E.E.; Okafor, P.C.; Ekpe, U.J.; Ogbobe, O.: Effect of halide ions on the corrosion inhibition of aluminium in alkaline medium using polyvinyl alcohol. *J. Appl. Polym. Sci.* **103**, 2810–2816 (2007). <https://doi.org/10.1002/app.25446>
-

[Article Google Scholar](#)

55. Solomon, M.M.M.; Umoren, S.A.A.; Udosoro, I.I.I.; Udoh, A.P.P.: Inhibitive and adsorption behaviour of carboxymethyl cellulose on mild steel corrosion in sulphuric acid solution. *Corros. Sci.* **52**, 1317–1325 (2010). <https://doi.org/10.1016/j.corsci.2009.11.041>
-

[Article Google Scholar](#)

56.Sangeetha, M.; Rajendran, S.; Sathiyabama, J.; Prabhakar, P.: Eco friendly extract of Banana peel as corrosion inhibitor for carbon steel in sea water. J. Nat. Prod. Plant Resour. **2**, 601–610 (2012)

[Google Scholar](#)

57.Wang, H.; Gao, M.; Guo, Y.; Yang, Y.; Hu, R.: A natural extract of tobacco rob as scale and corrosion inhibitor in artificial seawater. Desalination **398**, 198–207 (2016). <https://doi.org/10.1016/j.desal.2016.07.035>

[Article Google Scholar](#)

58.Sribharathy, V.; Rajendran, S.: Cuminum cyminum extracts as eco-friendly corrosion inhibitor for mild steel in seawater. ISRN Corros. **2013**, 1–7 (2013). <https://doi.org/10.1155/2013/370802>

[Article Google Scholar](#)

59.Hart Kalada, G.; Orubite-Okorosaye, K.; James Abodsede, O.: Corrosion inhibition of mild steel in simulated seawater by *Nymphae Pubscens* leaf extracts (NLE). Int. J. Adv. Res. Chem. Sci. **4**, 32–40 (2017). <https://doi.org/10.20431/2349-0403.0412004>

[Article Google Scholar](#)

60.Johnsirani, V.; Sathiyabama, J.; Rajendran, S.; Lydia Christy, S.M.; Jeyasundari, J.: The effect of *Eclipta alba* leaves extract on the corrosion inhibition process of carbon steel in sea water. Port. Electrochim. Acta. **31**, 95–106 (2013). <https://doi.org/10.4152/pea.201302095>

[Article Google Scholar](#)

61.Mobin, M.; Rizvi, M.; Olasunkanmi, L.O.; Ebenso, E.E.: Biopolymer from Tragacanth Gum as a green corrosion inhibitor for carbon steel in 1 M

HCl solution. ACS Omega **2**, 3997–4008
(2017). <https://doi.org/10.1021/acsomega.7b00436>

[Article Google Scholar](#)

62. Karthik, R.; Muthukrishnan, P.; Chen, S.M.; Jeyaprabha, B.; Prakash, P.: Anti-corrosion inhibition of mild steel in 1M hydrochloric acid solution by using *Tiliacora accuminata* leaves extract. Int. J. Electrochem. Sci. **10**, 3707–3725 (2014)
-

[Google Scholar](#)

63. Lerner, S.: Europeans Aim to Phase Out Toxic PFAS Chemicals by 2030., <https://theintercept.com/2019/12/19/pfas-chemicals-europe-phase-out/>
64. Finšgar, M.; Jackson, J.: Application of corrosion inhibitors for steels in acidic media for the oil and gas industry: a review. Corros. Sci. **86**, 17–41 (2014). <https://doi.org/10.1016/j.corsci.2014.04.044>
65. Altemimi, A.; Lakhssassi, N.; Baharlouei, A.; Watson, D.G.; Lightfoot, D.A.: Phytochemicals: extraction, isolation, and identification of bioactive compounds from plant extracts. Plants **6**, 1–23 (2017). <https://doi.org/10.3390/plants6040042>
-

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Enabulele, D.O., Bamigboye, G.O., Solomon, M.M. *et al.* Exploration of the Corrosion Inhibition Potential of Cashew Nutshell on Thermo-Mechanically Treated Steel in Seawater. *Arab J Sci Eng* **48**, 223–237 (2023).

<https://doi.org/10.1007/s13369-022-06981-5>

[Download citation](#)

- Received 16 December 2021
- Accepted 15 May 2022
- Published 09 June 2022
- Issue Date January 2023
- DOI <https://doi.org/10.1007/s13369-022-06981-5>

Keywords